

INTERNATIONAL SYMPOSIUM AND WORKSHOP ON ASTROCHEMISTRY



Understanding extraterrestrial molecular complexity
through experiments and observations

Non-thermal ion desorption from nitrile-bearing astrophysical ice analogues studied by electron and heavy ion bombardment

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Departamento
de Física



instituto de química

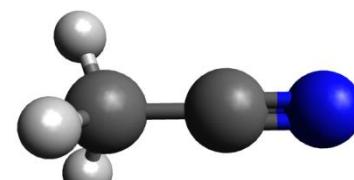
Universidade Federal do Rio de Janeiro



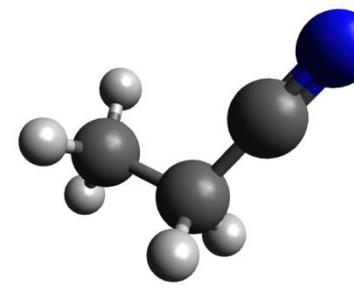
2 atoms	3 atoms	4 atoms	5 atoms	6 atoms	7 atoms	8 atoms	9 atoms	10 atoms	11 atoms	12 atoms	>12 atoms
H ₂	C ₃ *	c-C ₃ H	C ₅ *	C ₆ H	C ₆ H	CH ₂ C ₃ N HCO)OCH ₃	CH ₂ C ₄ H	CH ₂ C ₅ N	CH ₂ C ₆ N	c-C ₆ H ₆ *	HC ₁ N
AlF	C ₂ H	/C ₂ H	C ₄ H	/H ₂ C ₄	CH ₂ CHCN		CH ₂ CH ₂ CN	(CH ₃) ₂ CO	CH ₃ C ₆ H	n-C ₆ H ₁₂ CN	C ₆₀ *
AlCl	C ₂ O	C ₃ N	C ₄ Si	C ₂ H ₄ *	CH ₃ C ₂ H	CH ₂ COOH	(CH ₂) ₂ O	(CH ₂ OH) ₂	C ₂ H ₅ OCHO	C ₆ H ₇ CN	C ₇₀ *
C ₂ **	C ₂ S	C ₃ O	/C ₃ H ₂	CH ₂ CN	HC ₅ N	C ₇ H	CH ₃ CH ₂ OH	CH ₂ CH ₂ CH	CH ₂ OC(O)	C ₂ H ₅ OCH ₃	C ₆₀ **
CH	CH ₂	C ₃ S	c-C ₃ H ₂	CH ₃ NC	CH ₂ CHO	C ₆ H ₂	HC ₇ N		CH ₂ CHCH ₂		
	CH ⁺	HCN	C ₂ H ₂ *	H ₂ CCN	CH ₃ OH	CH ₃ NH ₂	CH ₂ OHCH	C ₈ H			
	CN	HCO	NH ₃	CH ₄ *	CH ₂ SH	c-C ₂ H ₄ O	/H ₂ C ₆ H*	CH ₂ C(O)N			
CO	HCO*	HCCN	HC ₃ N	HC ₃ NH ⁺	H ₂ CCOH	CH ₂ CHCH	C ₈ H-				
CO ⁺	HCS*	HCNH ⁺	HC ₂ NC	HC ₂ CHO	C ₆ H-	CH ₂ CCHC	C ₃ H ₃				
CP	HOC ⁺	HNCO	HCOOH	NH ₂ CHO	CH ₃ NCO	H ₂ NCH ₂ CN	H ₃ CH ₂ SH				
SiC	H ₂ O	HNCS	H ₂ CNH	C ₂ N		CH ₂ CHNH	(?)				
HCl	H ₂ S	HOCO	H ₂ O ₂	H ₂ O ₂							
KCl	HNC	H ₂ CO	H ₂ NCN	HNC ₃							
NH	HNO	H ₂ CN		HNC ₃							
NO	MgCN	H ₂ CS		SiH ₄ *	H ₂ CCNH	(?)					
NS	MgNC	H ₂ O ⁺	H ₂ COH ⁺	C ₂ N-	HNCN						
NaCl	N ₂ H ⁺	c-SiC ₃	C ₂ H-	HNCN							
OH	N ₂ O	CH ₃ *	HO ₂ CN								
PN	NaCN	C ₃ N-	HNCNH								
SO	OCS	PH ₃	CH ₂ O								
SO ⁺	SO ₂	HCNO	NH ₄ ⁺								
SiN	c-SiC ₂	HOCN	H ₂ NCO ^(?)	NCCNH ⁺							
SiO	CO ₂ *	HSCN									
SiS	NH ₂	H ₂ O ₂									
CS	H ₃ [±] (*)	C ₂ H ₄									
HF	SICN	HMgNC									
HD	AINC	HCCO									
FeO?	SINC										
O ₂	HCP										
CF ⁺	CCP										
SiH?	AOH										
PO	H ₂ O ⁺										
AlO	H ₂ Cl ⁺										
OH ⁺	KCN										
CN-	FeCN										
SH ⁺	HO ₂										
SH	TiO ₂										
HCl ⁺	C.N										
TiO	Si ₂ C										
	2015										
ArH ⁺											
NO?											

Several organic and inorganic nitriles and isonitriles

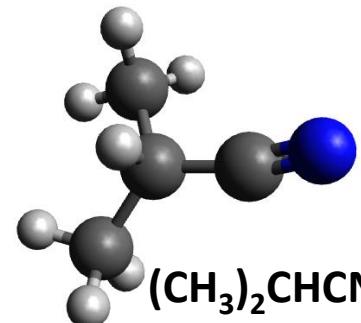
<https://www.astro.uni-koeln.de/cdms/molecules>



CH₃CN



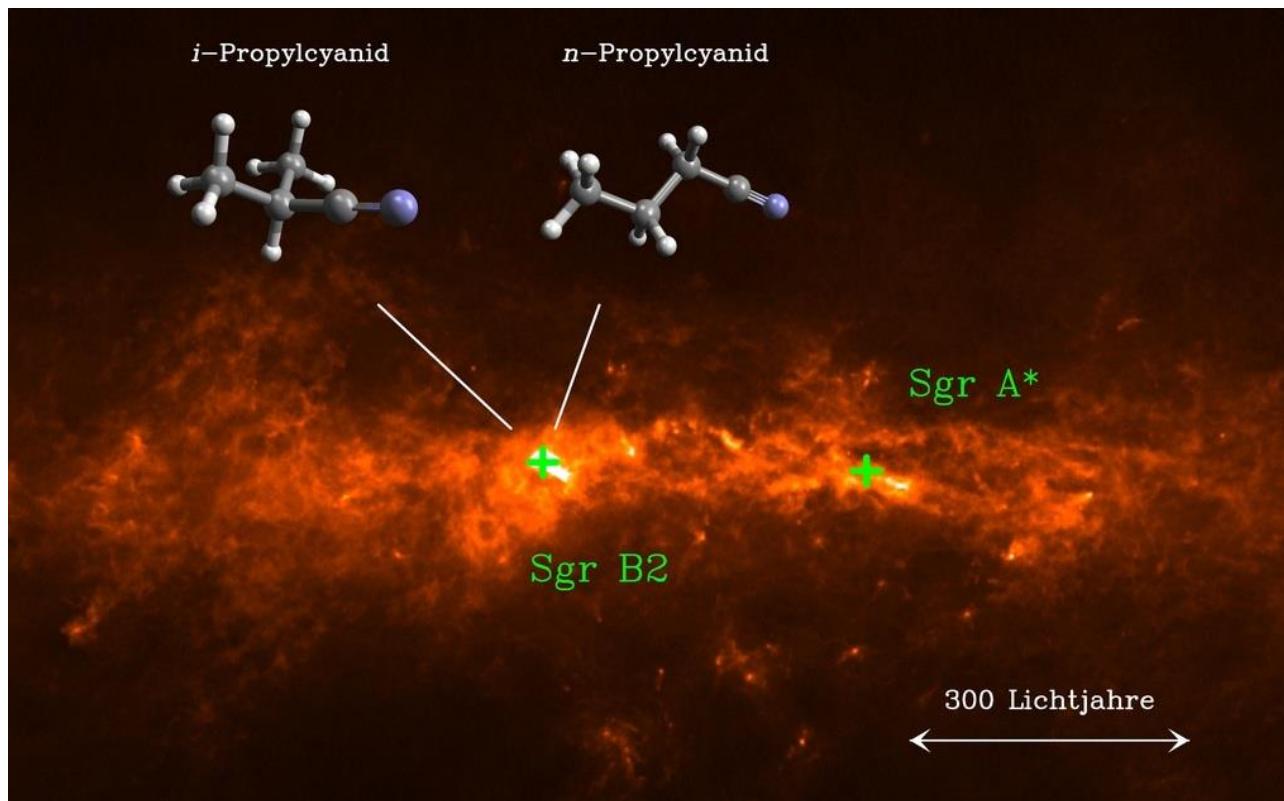
CH₃CH₂CN



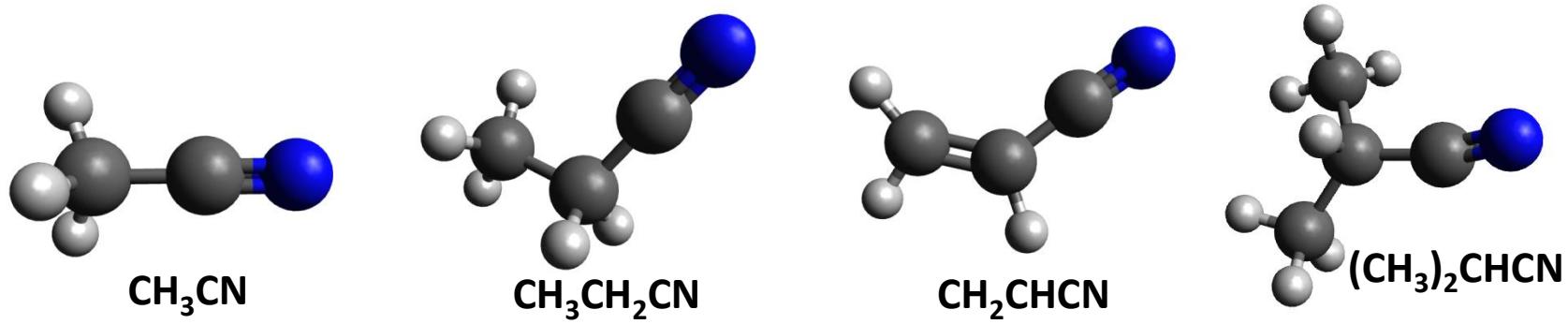
(CH₃)₂CHCN

Relevant Interstellar Nitriles

Increasing complexity of the organic —C≡N series



Belloche et al. *Science*, 2014 , 345, 1584-1587



Relevant Interstellar Nitriles

- Very abundant in space (very common in star forming regions)
- Important in the formation of amino acids;
- CH₃CN is a good probe to estimate **temperature** and **column densities** based on observations of a single rotational transition.
- Tracer for Hot Molecular Cores (**HMCs**)
- Enhanced abundance of CH₃CN in **warm** (T = 100–300 K) and **dense** (n_{H₂}= 10⁶ – 10⁸ cm⁻³) environments;

Solar System bodies

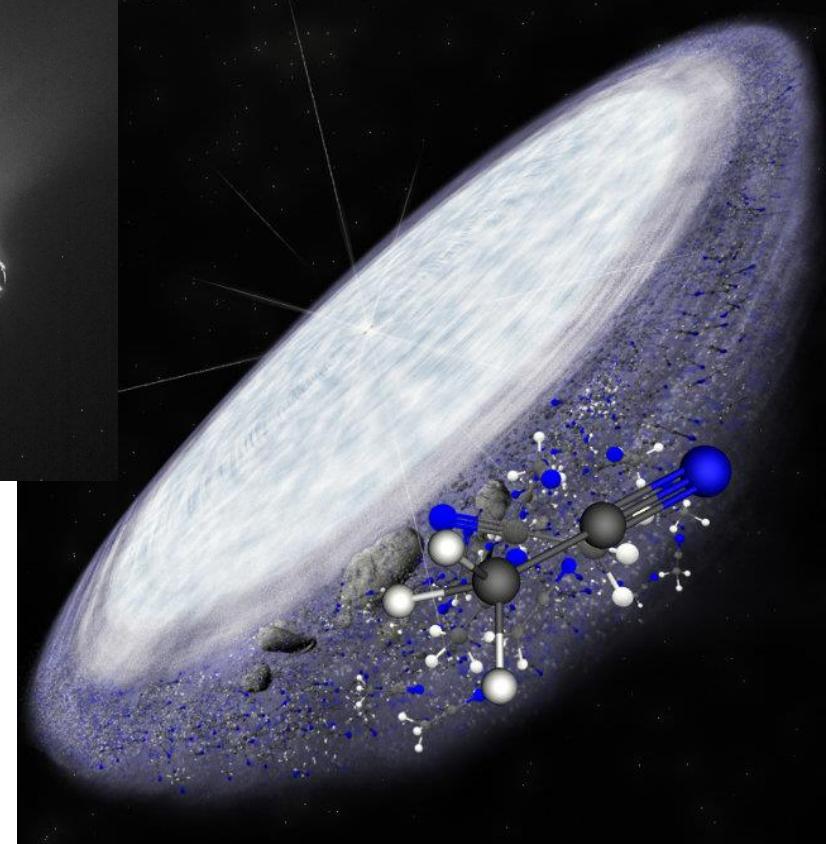
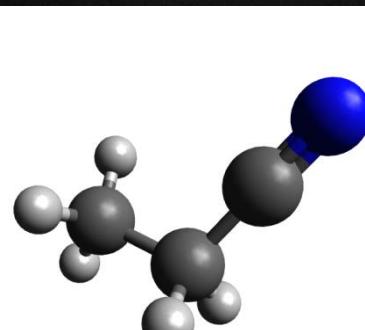
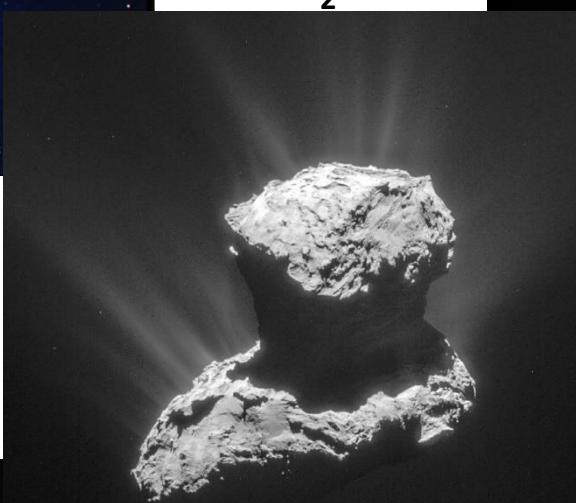
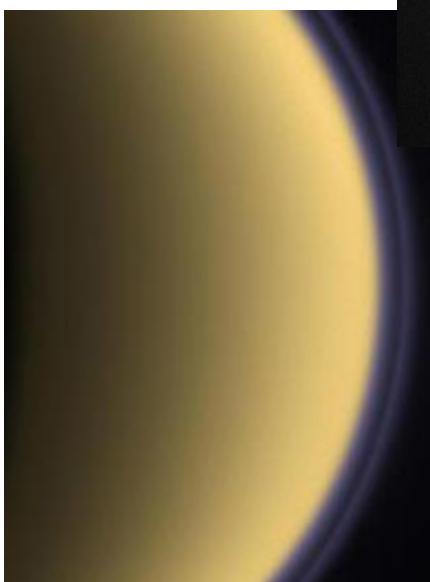
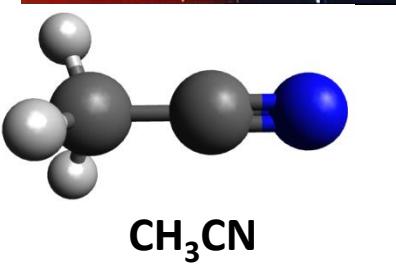
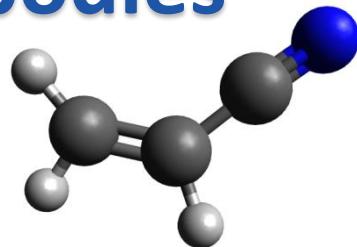


Image credit: B. Saxton / NRAO / AUI / NSF.

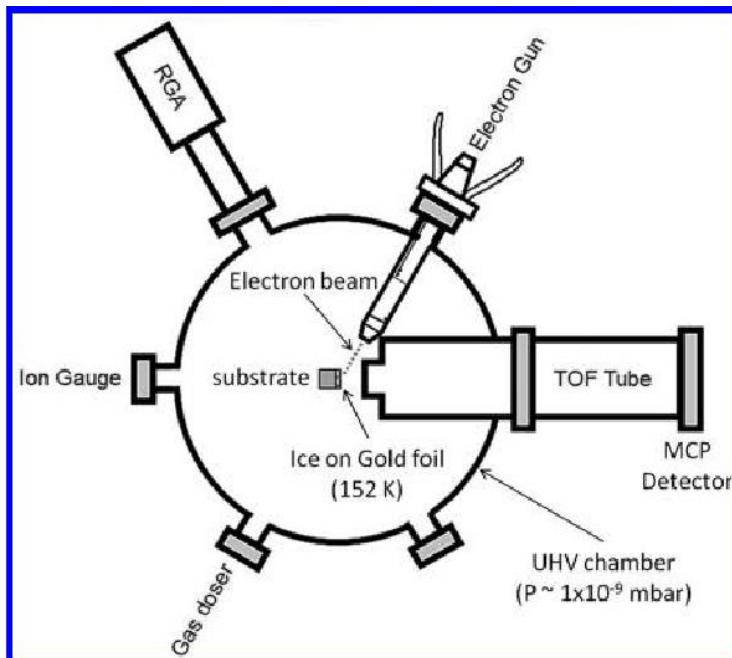
Nitrile Chemistry

Problems:

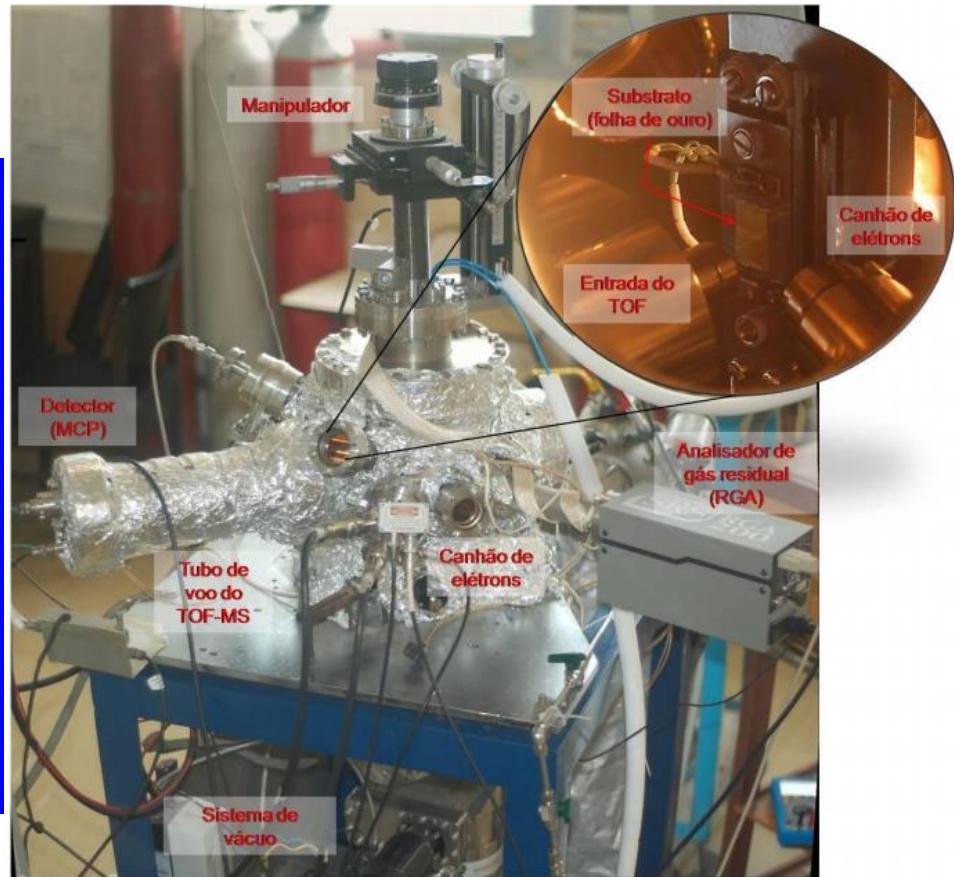
- How such **complex nitriles may be formed?**
- Not enough complex species can be produced in the **gas phase by known reaction routes;**
- What is the role played by **dust grains/ ice mantles?**
- What is the influence of **ionizing radiation?**
- Does **ion desorption influence gas abundances?**
- Is their **chemistry connected?**
- Is the same chemistry happening in other sources?

Laboratory work

- Surface processes are poorly known;
- Surface Science Techniques under conditions that resemble those found in the ISM;
- Non-thermal desorption processes:



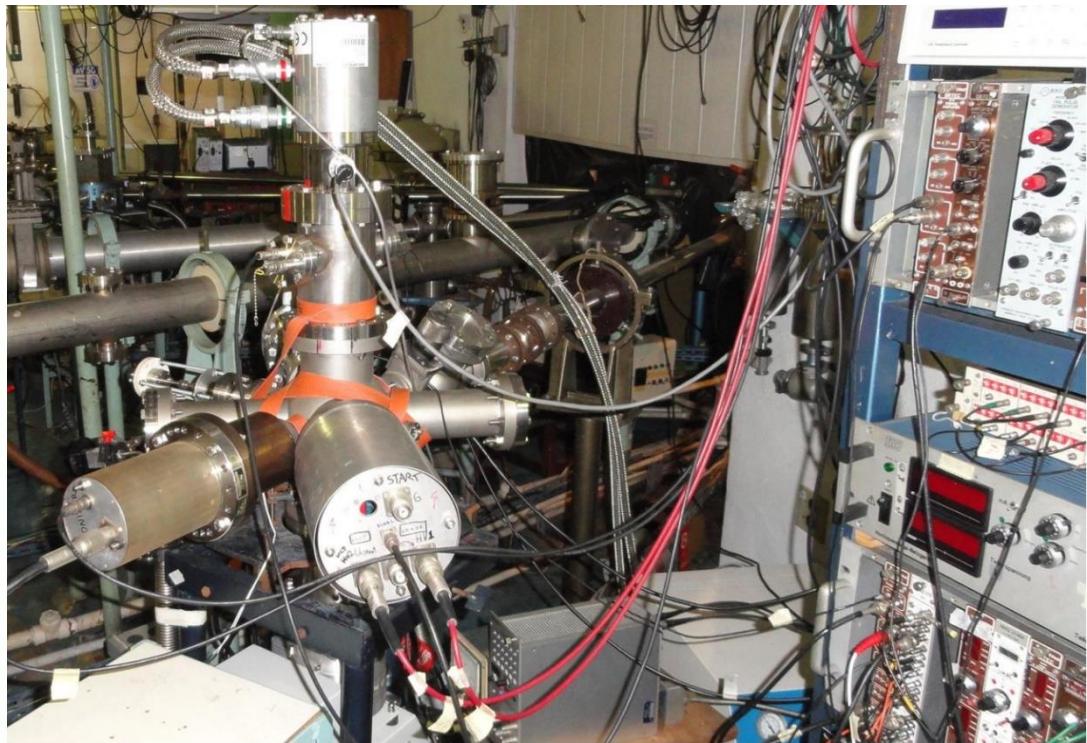
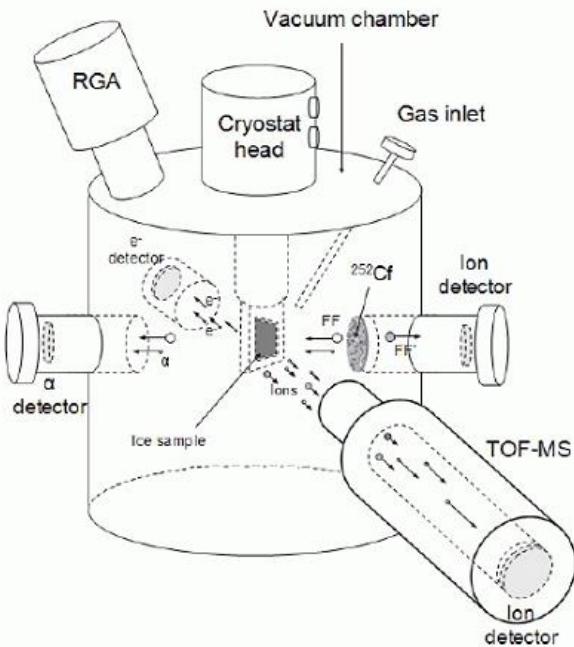
electrons in/ ions out



Electron Stimulated Ion Desorption (ESID)

Laboratory work

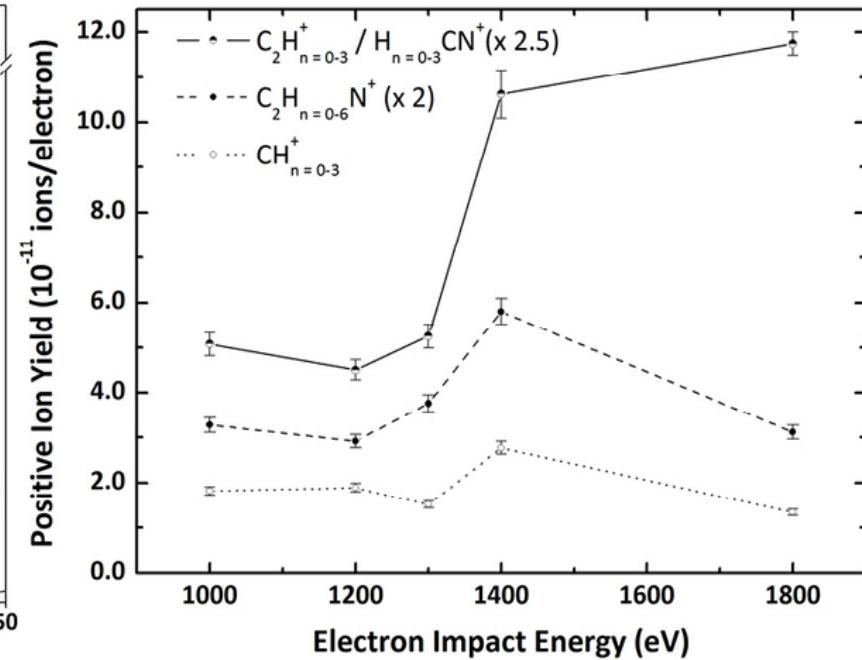
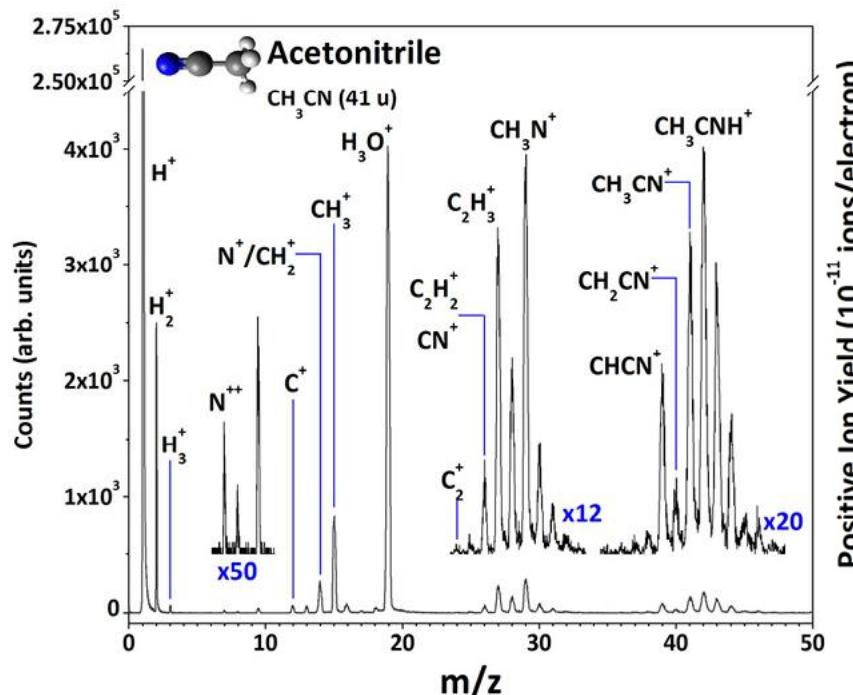
- Surface processes are poorly known;
- Surface Science Techniques under conditions that resemble those found in the ISM;
- Non-thermal desorption processes:



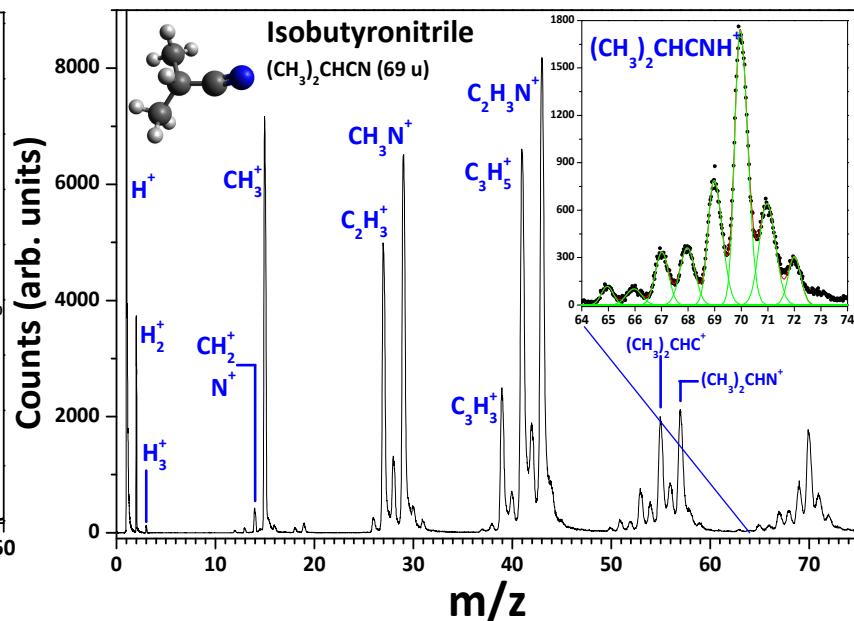
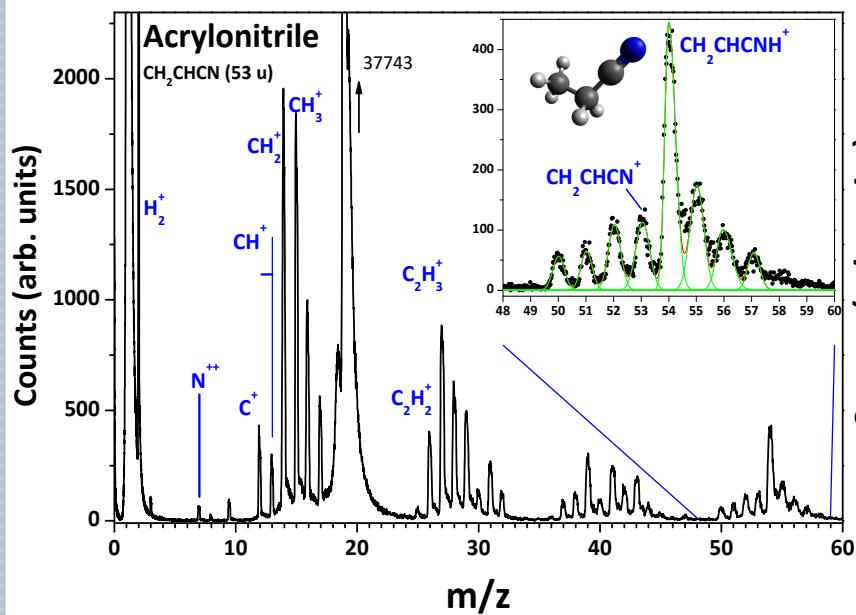
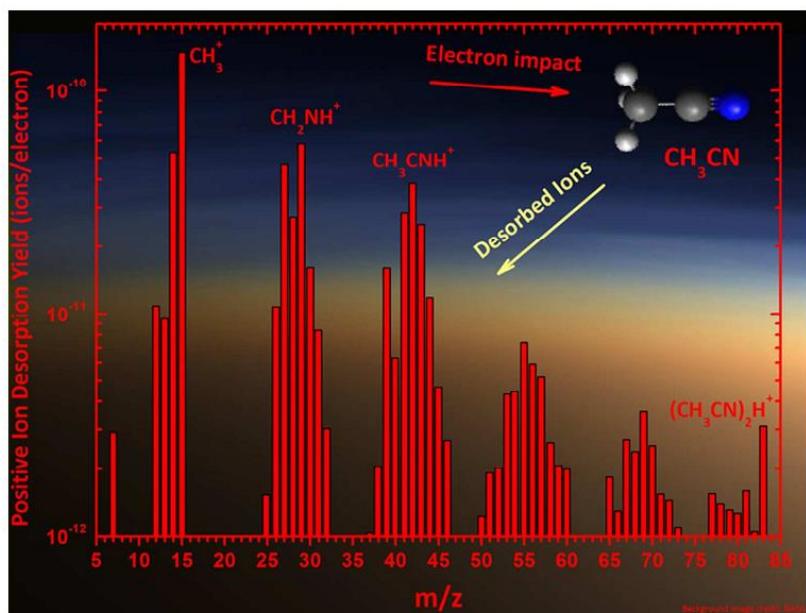
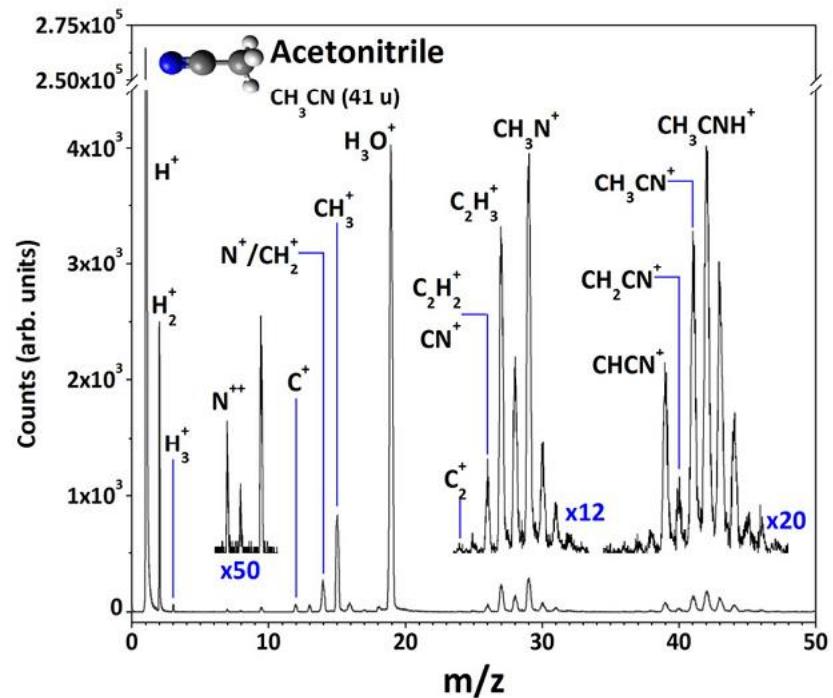
ion in/ secondary ions out

Plasma Desorption Mass Spectrometry (PDMS)

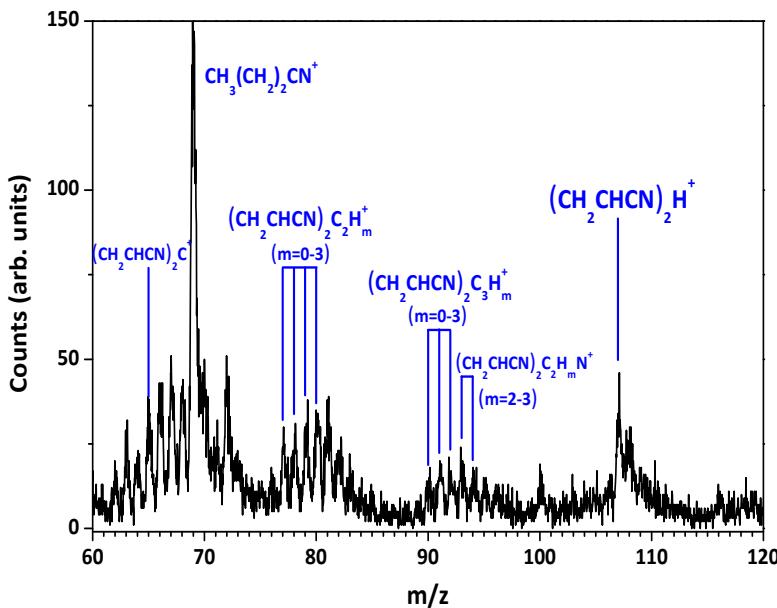
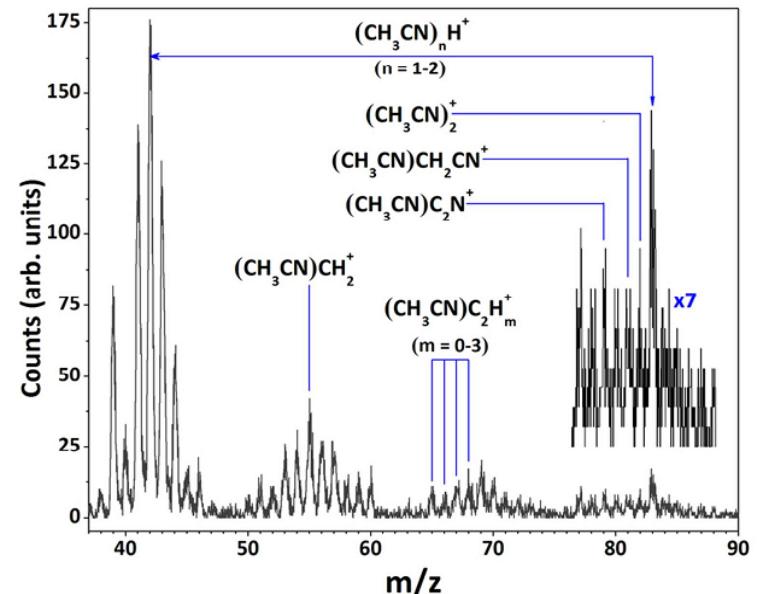
Laboratory experiments - ESID



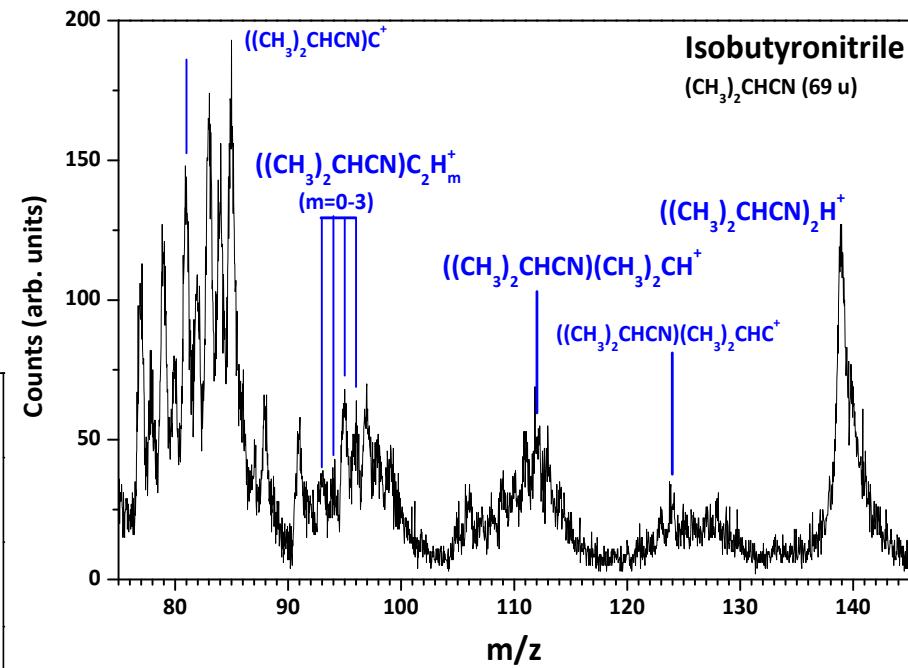
Ion desorption increases at 1400 eV, which is 3.5 times the ionization threshold for the CH_3CN N1s core level at 406 eV.



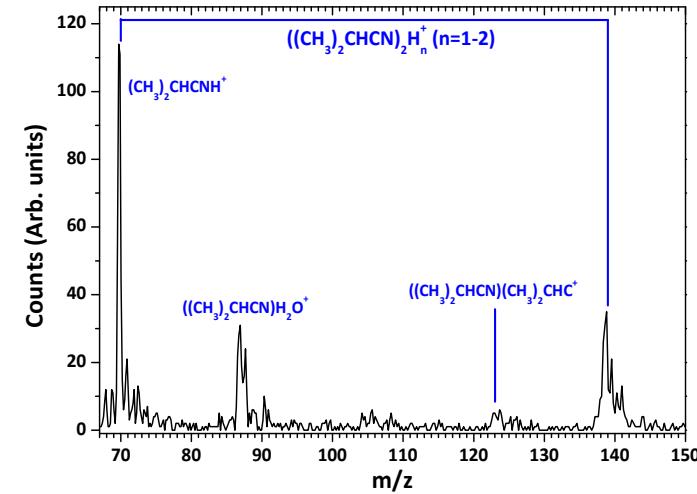
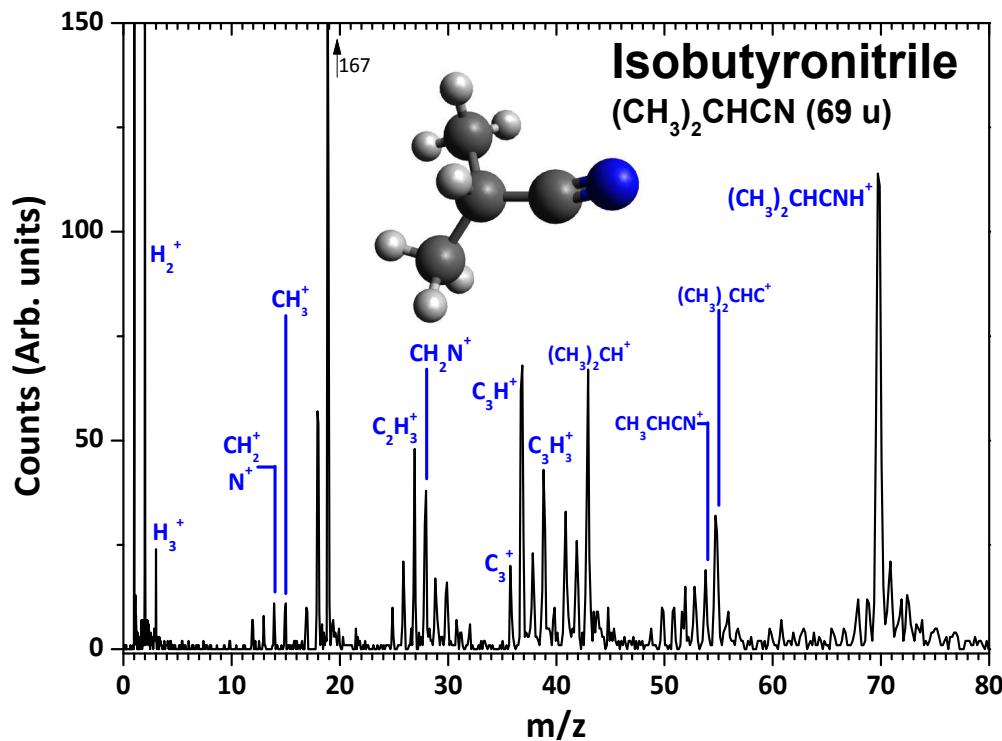
Laboratory experiments - ESID



Desorption of ion clusters by electron impact (2.3 keV)



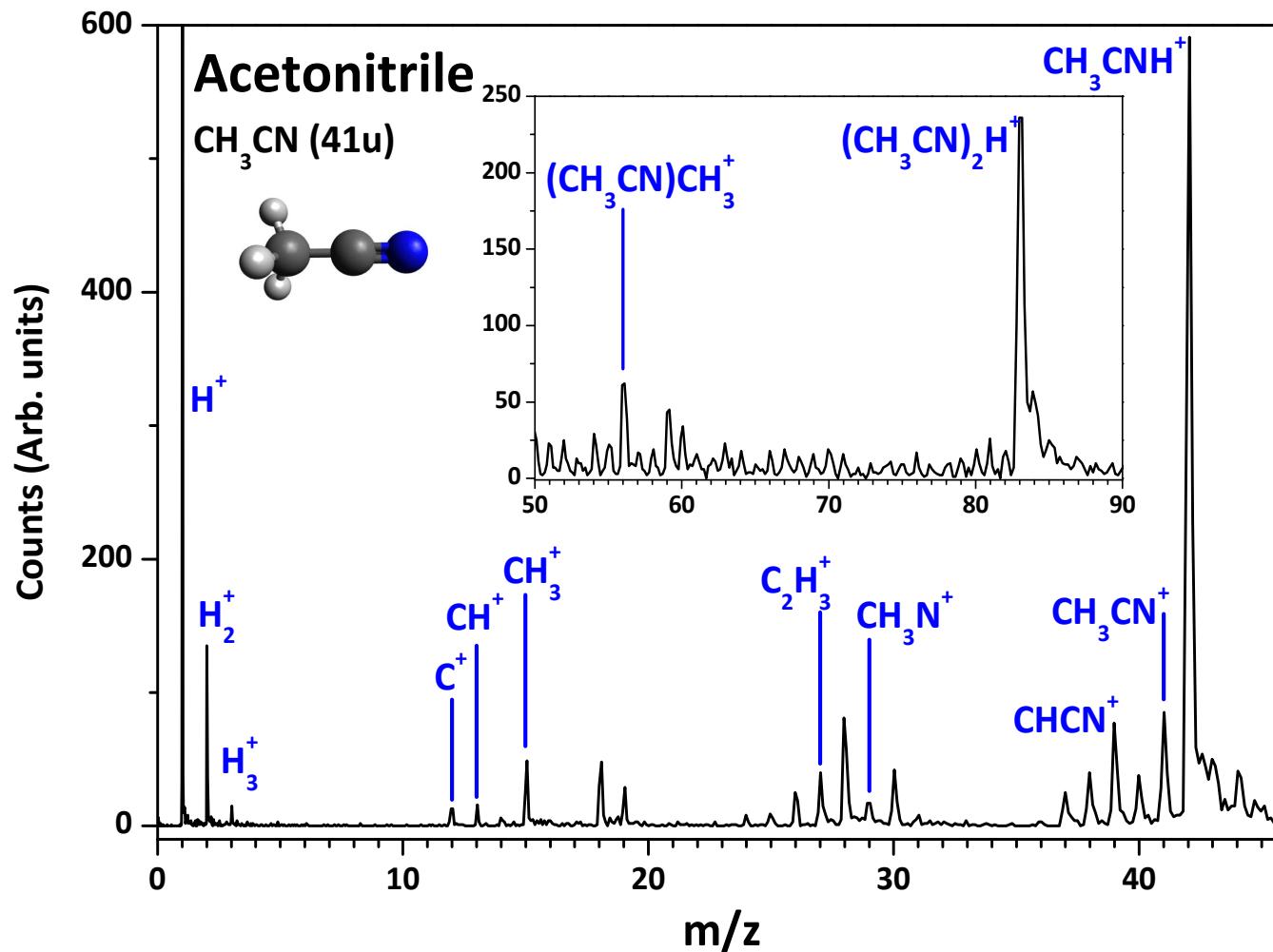
Laboratory experiments - PDMS



Desorption of $(\text{CH}_3)_2\text{CHCN}$ ion clusters

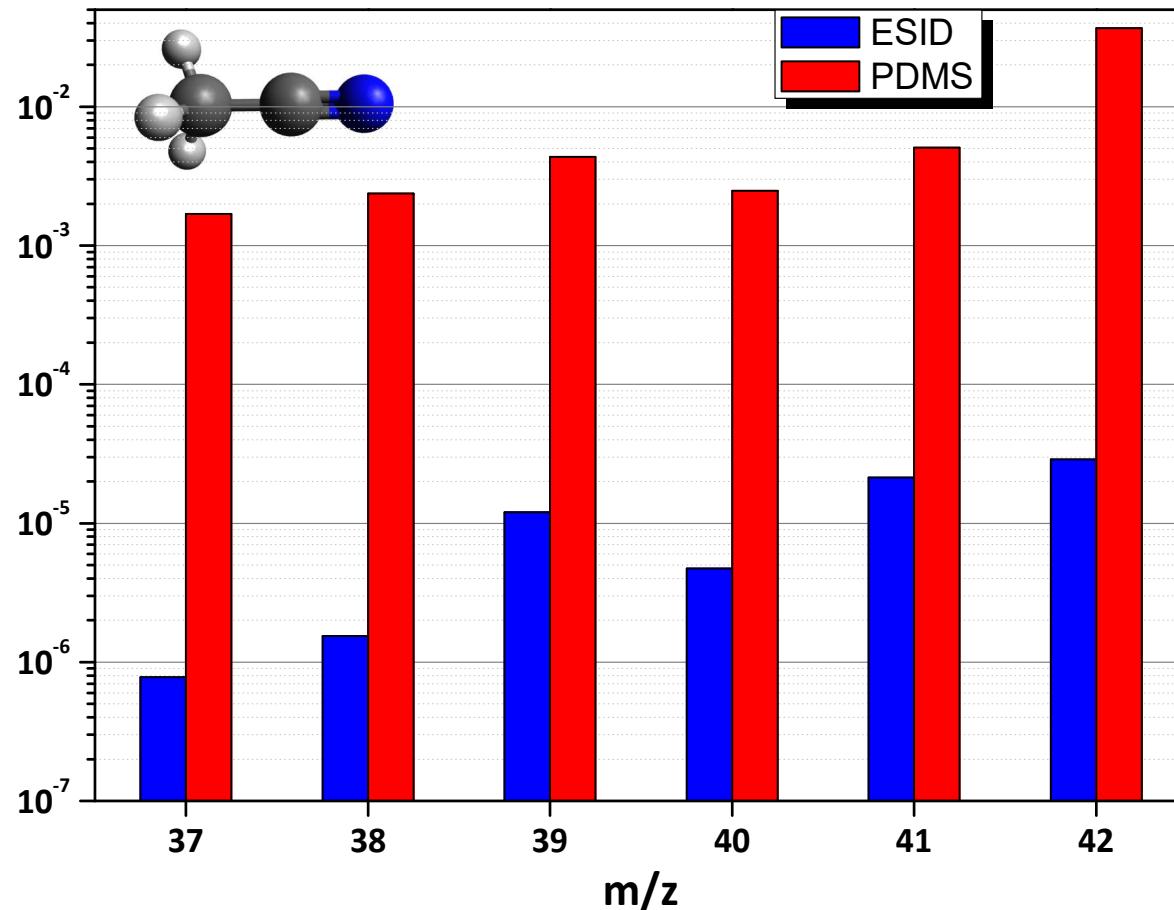
PDMS mass spectrum of $(\text{CH}_3)_2\text{CHCN}$ at 100 K

Laboratory experiments - PDMS



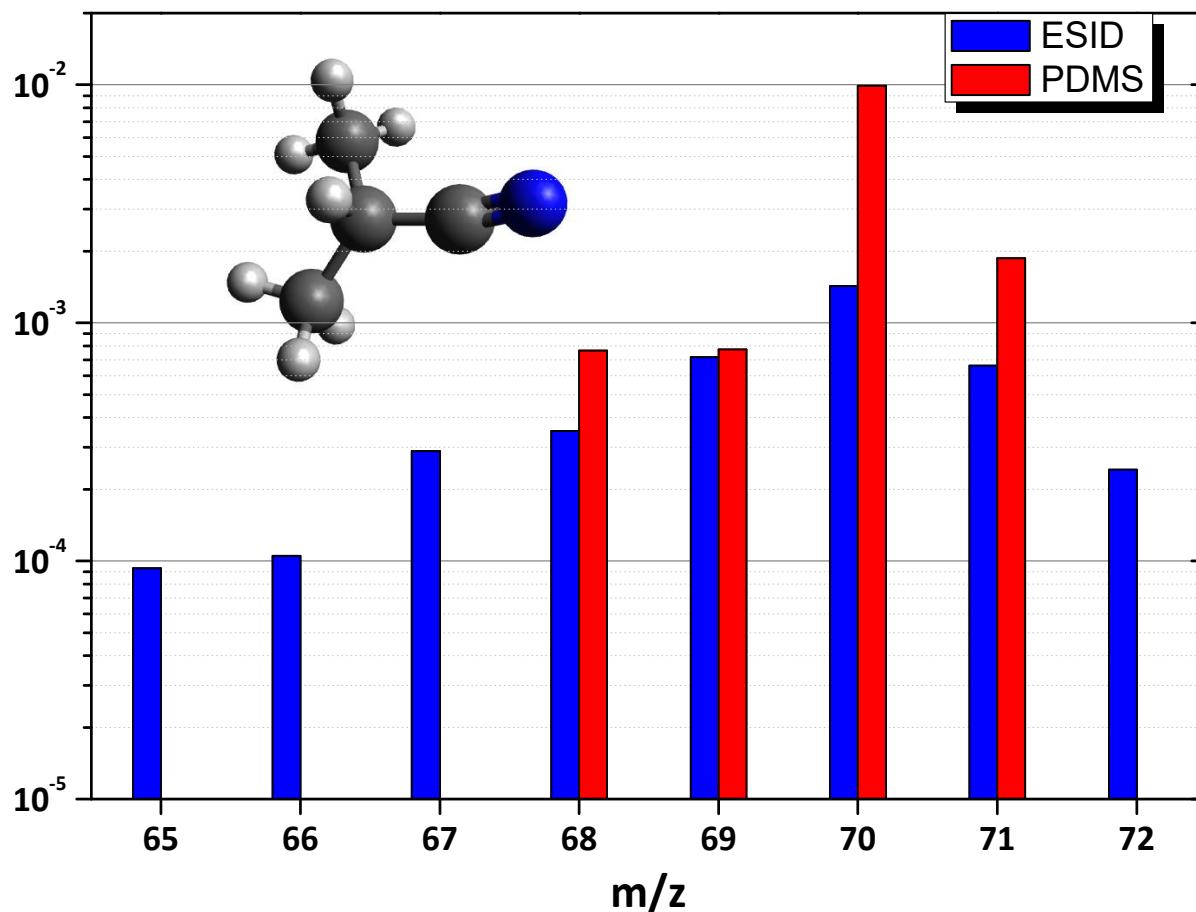
PDMS mass spectrum of CH_3CN at 100 K. Inset: Desorption of CH_3CN ion clusters

ESID and PDMS comparison



- Stronger fragmentation in **ESID**
- **Proton transfer** processes during ion desorption

ESID and PDMS comparison



- Stronger fragmentation on surface in **ESID** in respect to CH_3CN ;
- Similar Ion Yield for $(CH_3)_2CHCN^+$ ($m/z = 69$) ion desorption in ESID and PDMS;
- **Proton transfer** processes during ion desorption.

Summary Remarks

- Strong fragmentation on surface and ion desorption is observed for all studied nitriles;
- Fragmentation caused by electrons is initiated by Coulomb explosion after Auger electronic decay;
- Predominance for saturated and protonated fragments desorption. The last might play a role in ion-neutral reactions on gas-phase;
- Cluster ion desorption may be a route for delivering for complex molecules (nitriles) to the cold interstellar and circumstellar material exposed to ionizing radiation
- Similar conclusions can be ascribed to the Titan atmosphere,
 - where a set of complex nitriles is known to exist

Acknowledgements

Thank you for your attention!

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